

Enzymatic Resolution of *O*-(Methoxymethyl)-Protected Tropane-diols

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Dedicated to Professor *Gerhard Simchen* on the occasion of his 75th birthday

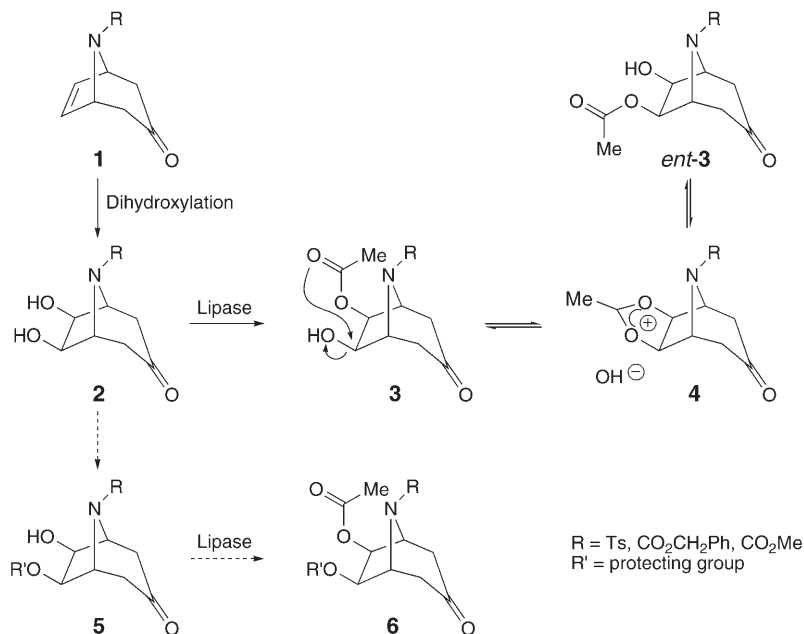
A convenient synthetic route to enantiomerically pure tropane-diol building blocks is described. The reaction sequence started from tropenone derivatives **1**, which were dihydroxylated to give 6,7-dihydroxytropanone derivatives **2**. After introduction of the methoxymethyl (MOM) protecting group in diol **2a**, a lipase-mediated resolution of the resulting racemic mono-MOM ether (\pm)-**5d** with vinyl acetate and vinyl trifluoroacetate gave the acetates (–)-**6d** and (–)-**6f**, respectively, with 96–99% ee, and MOM ether (+)-**5d** with up to 89% ee. Deacetylation of (–)-**6d** afforded quantitatively MOM ether (–)-**5d** with 99% ee, the absolute configuration of which was assigned *via* the modified *Mosher* method to be (*R*) at C(6). Enzymatic treatment of unprotected diol **2a** with vinyl trifluoroacetate or alkoxycarbonylation resulted in the formation of *C_s*-symmetrical products **9** and **12** rather than the desired desymmetrized derivatives.

Introduction. – Due to their biological activities, tropane alkaloids and derivatives thereof have been studied extensively over the last decades [1] (for some recent examples, see [2]). Furthermore, tropanes (= 8-methyl-8-azabicyclo[3.2.1]octanes) also provide useful chiral scaffolds for ligands in asymmetric catalysis [2e][3]. While the majority of tropane syntheses relied on either *de novo* routes or scopolamine as starting material [1][2], little work has been devoted to the functionalization *via* enantioselective deprotonation of tropinone [4], hydroboration of tropenone derivatives **1** to the corresponding chiral alcohol [5], or enzymatic resolution of the latter [6]. In a previous paper, we reported the dihydroxylation of tropenone derivatives **1** and subsequent lipase-mediated acetylation of the corresponding diols **2** to give the mono acetates **3**, which might be further functionalized (*Scheme 1*)¹⁾ [8]. However, it turned out that the direct enzymatic acylation of the diols **2** provided racemic mixtures of mono acetates **3** [9].

We anticipated that, even if the enzyme shows some enantioselectivity, the presence of the free OH group in compounds **3** accelerates the formation of an acetoxonium ion **4**, which may be opened to yield either **3** or *ent*-**3** (*Scheme 1*). Such acyl migration is well-known in carbohydrate chemistry [10], terminal 1,2-diols [11], mono- and diglycerides [7][12], 1,3-diols [13], and amino alcohols [14], resulting in reduced regio- and enantioselectivities. *Bäckvall* and co-workers utilized fast acyl migrations for dynamic kinetic asymmetric transformations, *i.e.*, one-pot lipase-catalyzed acylations of

¹⁾ For lipase-catalyzed reactions of 1,2-diols, see [7].

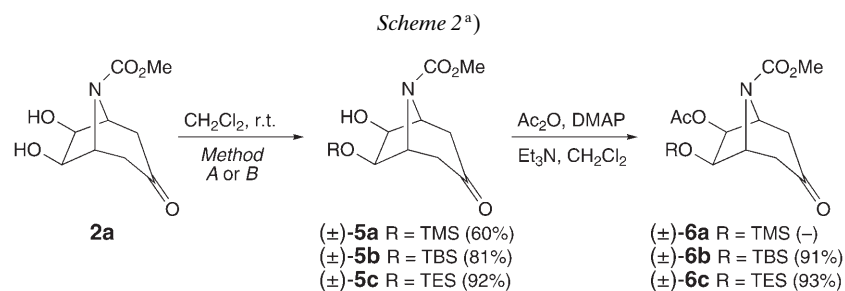
Scheme 1



1,2- and 1,3-diols, Ru-catalyzed epimerizations, and intramolecular acyl migration to affect the formation of optically pure *syn*-1,2- and *syn*-1,3-diacetates [15]. With regard to 6,7-dihydroxytropane derivatives **2**, neither of the above mentioned approaches was successful, and, thus, we decided to introduce a protecting group prior to enzymatic acetylation (Scheme 1). The results along this direction are reported below.

Results and Discussion. – To find a suitable protecting group, dihydroxy-tropane-carboxylate **2a**, which was accessible in 75% yield by treatment of **1** (R = CO₂Me) with K₂OsO₄ and *N*-methylmorpholine *N*-oxide monohydrate (NMO) in a mixture of acetone/H₂O/*t*-BuOH 10 : 2 : 1 [9][16], was first reacted with various silylation reagents (Scheme 2). While treatment with TMSCl in the presence of *Hünig* base yielded the corresponding racemic mono-silylated carboxylate **5a** in 60% (Method A), both silyloxy derivatives **5b** and **5c** were obtained in a similar manner in 81 and 92% yield, respectively (Method B). In the case of **5b**, the corresponding disilylated compound was obtained in 3% yield. The chemical acetylation of silyloxy derivatives (±)-**5b** and **5c** with Ac₂O, DMAP, and Et₃N in CH₂Cl₂ gave the corresponding acetates (±)-**6b** and **6c**, respectively in > 90% yield (Scheme 2). However, under these conditions the TMS group in (±)-**5a** was removed, finally giving the corresponding mono- and diacetate. It must be noted that desilylation of TMS derivative **5a** was observed even at storage.

Treatment of protected dihydroxy compounds (±)-**5** with vinyl acetate and lipases Chirazyme L-1 and L-6 and Novozyme 435 (*Candida antarctica* lipase B), however, did not give any trace of the desired acylation products **6**. Presumably, the bulky silyl

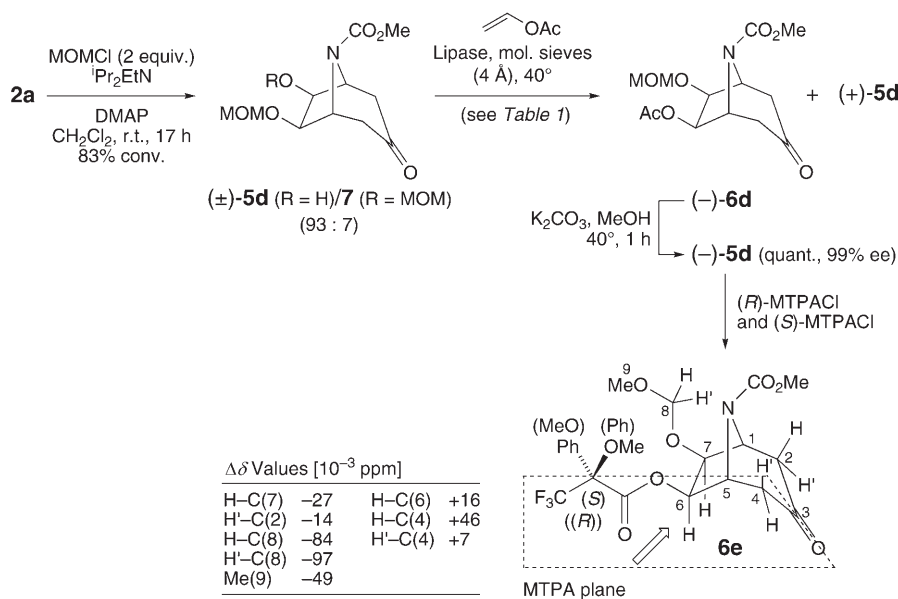


^{a)} *Method A*: Me_3SiCl (TMSCl), Et_3NPr_2 , 1 h. *Method B*: $(t\text{-Bu})\text{Me}_2\text{SiCl}$ (TBSCl) or Et_3SiCl (TESCl), Et_3N , 4-(dimethylamino)pyridine (DMAP), 5 h.

groups prevent access to the active site of the enzymes. Even Chirazyme L-5 (*Candida antarctica* lipase A), which is known to accept sterically hindered alcohols [17], did not yield **6**.

Therefore, methoxymethyl (MOM) was introduced as an alternative protecting group (Scheme 3). The lipophilic binding site of the lipases might favor a MOM acetal over more polar carboxylates, while simultaneously allowing the OH group to be attacked by the enzyme. Treatment of the dihydroxy derivative **2a** with 2 equiv. MOMCl in the presence of Hünig base and DMAP gave a mixture of mono- and bis-MOM ether **5d** and **7** in a ratio of 93:7, which was not separable by column chromatography (Scheme 3). The chemical acetylation of **5d** to **6d** under the above

Scheme 3. MOM Protection of Diol **2a**, Subsequent Lipase-Catalyzed Resolution of $(\pm)\text{-5d}$, and Assignment of the Absolute Configuration of $(-)\text{-5d}$ According to Kakisawa and co-workers [18]



mentioned conditions proceeded in 74% yield. For enzymatic resolution of (\pm)-**5d** with vinyl acetate, various lipases and solvents were investigated at 40° (Table 1).

Table 1. Enzymatic Acetylation of MOM Ether (\pm)-**5d** with Various Lipases to Acetate **6d**

Entry	Solvent	Enzyme ^{a)}	Time [h]	Conv. [%]	Yield [%] ^{b)}	% ee ^{c)}
1	Toluene	Chirazyme L-1	47	23	–	5
2	Toluene	Chirazyme L-5	48	3	–	–
3	Toluene	Chirazyme L-6	47	38	33	96
4	Toluene	Novozyme 435	48	7	–	–
5	CH ₂ Cl ₂	Chirazyme L-1	26	5	–	–
6	CH ₂ Cl ₂	Chirazyme L-5	24	3	–	–
7	CH ₂ Cl ₂	Chirazyme L-6	24	2	–	–
8	CH ₂ Cl ₂	Novozyme 435	26	3	–	–
9	Et ₂ O	Chirazyme L-1	4.5	14	–	50 ^{d)}
10	Et ₂ O	Chirazyme L-1	16	38	–	15 ^{d)}
11	Et ₂ O	Chirazyme L-1	18	51	41	8 ^{d)}
12	Et ₂ O	Chirazyme L-5	24	–	–	–
13	Et ₂ O	Chirazyme L-6	4.0	48	38	96
14	Et ₂ O	Chirazyme L-6	5.5	40 ^{e)}	31	96
15	Et ₂ O	Novozyme 435	6.0	11	–	> 99
16	Et ₂ O	Novozyme 435	24	51 ^{e)}	45	99

^{a)} Chirazyme L-1 from *Pseudomonas* sp., Chirazyme L-5 from *Candida antarctica*, Chirazyme L-6 from *Pseudomonas* cep., and Novozyme 435 from *Candida antarctica*. ^{b)} Yield of isolated **6d**. ^{c)} Enantioselectivities were determined by capillary GC on chiral stationary phase *Bondex un-β*. ^{d)} Opposite enantiomer (+)-**6d** is preferred. ^{e)} (+)-**5d** was isolated in 59% yield with 65% ee (Entry 14) and in 40% yield with 82% ee (Entry 16).

Lipase Chirazyme L-6 gave promising results (Entries 3 and 13). In toluene and Et₂O, respectively, good conversions and high enantiomeric excesses of 96% ee were obtained. In one case, the hydroxy derivative (+)-**5d** was isolated in 59% yield with 65% ee besides enantiomerically pure (–)-**6d** (96% ee; Entry 14). In Et₂O, also Novozyme 435 led to exceptional ee values of 99% but reacted much slower than Chirazyme L-6 (Entries 15 and 16). After 24 h reaction time, racemic MOM derivative (\pm)-**5d** was resolved to give (–)-**6d** with 99% ee, and (+)-**5d** in 40% yield and 82% ee (Entry 16). The other lipases Chirazyme L-1 and L-5 are less suitable independent of the solvent, giving low conversion and enantiomeric excess.

Whereas Et₂O was found to be the best solvent, CH₂Cl₂ was not suitable regardless of the enzyme used (Entries 5–8). Similar results were obtained for acetone and MeCN (not shown). It should be noted that, in Et₂O, Chirazyme L-1 resulted in the formation of the opposite enantiomer (+)-**6d** (Entries 9–11).

Concerning the further functionalization, the MOM acetate (–)-**6d** was deacetylated with K₂CO₃ in MeOH [18]. The reaction proceeded cleanly, and the enantiomerically pure MOM hydroxy derivative (–)-**5d** was obtained quantitatively with 99% ee (Scheme 3). The absolute configuration of (–)-**5d** was elucidated by NMR analysis according to the modified Mosher method [19] (Scheme 3). Compound (–)-**5d** was treated with (+)-(*S*)- and (–)-(*R*)-MTPACl (MTPACl = α -methoxy- α -(trifluoromethyl)phenylacetyl chloride) in the presence of pyridine at room temperature to give the

diastereoisomeric *Mosher* esters (*R*)-**6e** and (*S*)-**6e**. $^1\text{H-NMR}$ Spectra in CDCl_3 at room temperature, however, revealed a significant broadening of the signals for H–C(2), H–C(4), H–C(1), and H–C(5), rendering an assignment impossible. After considerable experimentation, $\text{C}_2\text{D}_2\text{Cl}_4$ at 393 K turned out to give the best results, thus avoiding any problems with rotamers at lower temperatures [9]. The $^1\text{H},^{13}\text{C}$ long-range COSY of ester (*R*)-**6e** at 393 K revealed 3J correlations between the ^{13}C signal of C(8) at δ 96.4 ppm and the ^1H signal of Me(9) at δ 3.30 ppm, as well as the *doublet* for H–C(7) at δ 4.14 ppm. The ^1H signal of the ester MeO group at δ 3.70 ppm showed a 3J coupling with the ^{13}C signal of the ester CO group. The ^{13}C signals of C(2) and C(4) at δ = 45.4 and 44.9 ppm were assigned according to 3J correlation of C(2) with H–C(7) at δ 4.14 ppm. In the $^1\text{H},^{13}\text{C}$ long-range COSY of the diastereoisomeric ester (*S*)-**6e** a 3J coupling between the C(4) signal at δ 44.9 ppm and the *doublet* of H–C(6) at δ 5.10 ppm, and a 3J coupling between the signal of C(2) at δ 45.5 ppm and the *doublet* of H–C(7) at δ 4.11 ppm were observed. Further assignments were accomplished by H,H-COSY measurements of both esters **6e** and the $\Delta\delta(\delta_S - \delta_R)$ values were obtained (Scheme 3). Applying *Kakisawa's* rules [19], the configuration at C(6) of (–)-**5d** was assigned as (*R*).

In contrast to the deacetylation, deprotection of (–)-**6d** resulted in all cases in the formation of racemic acetate (\pm)-**3a** (Scheme 4 and Table 2).

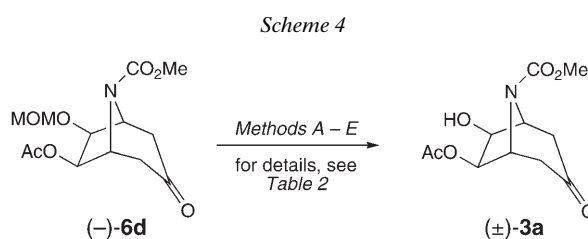


Table 2. Deprotection of MOM Acetate (–)-**6d** to Acetate **3a**

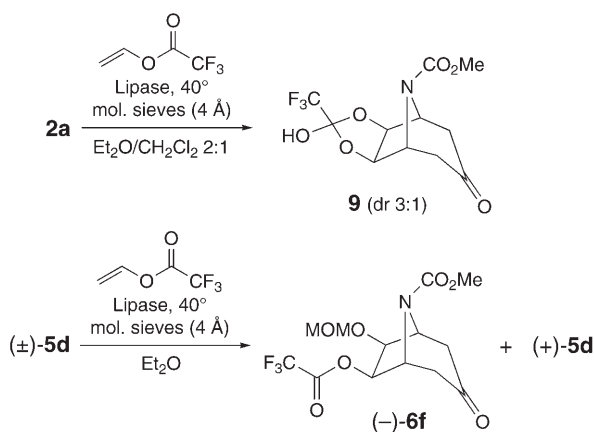
Method	Conditions	Conversion [%]
A	Me_3SiBr , CH_2Cl_2 , reflux, 24 h	36
B	$\text{BF}_3 \cdot \text{OEt}_2$, $\text{MeCN}/\text{H}_2\text{O}$, 2 h	50
C	TsOH , dioxane/ H_2O 6 : 1, reflux, 5 h	48
D	TsOH , dioxane/ H_2O 3 : 1, reflux, 6 h	73
E	TFA, CH_2Cl_2 , r.t., 5 h	quant. ^{a)}

^{a)} Yield of isolated (\pm)-**3a**: 66%.

As shown in Table 2, Me_3SiBr in refluxing CH_2Cl_2 [20], $\text{BF}_3 \cdot \text{OEt}_2$ in a mixture of $\text{MeCN}/\text{H}_2\text{O}$ [21] and catalytic amounts of TsOH in dioxane/ H_2O [22] led to low conversions. Upon increasing the amount of H_2O (Method D), the conversion was increased to 73%, but besides the desired acetate **3a**, 10% of dihydroxy derivative **2a** was formed. Complete removal of the MOM group was realized with TFA in CH_2Cl_2 at room temperature [23], and the target acetate **3a** was isolated in 66% yield, albeit being racemic (Method E).

Probably, as already observed for the enzymatic acylation of the dihydroxy carboxylate **2a** [9], owing to the free OH group again, the Ac migration dominated under the acidic reaction conditions. We anticipated that the electron-withdrawing CF₃CO group might disfavor the formation of the intermediate acetoxonium cations **4**. Surprisingly, little is known about the enzymatic resolution with lipase employing vinyl trifluoroacetate except one publication by *Miyazawa et al.* who studied the *Achromobacter* sp. lipase [24]. We used Chirazyme L-6 and Novozyme 435 for the enzymatic acylation of **2a** with vinyl trifluoroacetate at 40°. The solvent mixture Et₂O/CH₂Cl₂ 2 : 1 was chosen, because **2a** was not soluble in pure Et₂O (*Scheme 5*).

Scheme 5



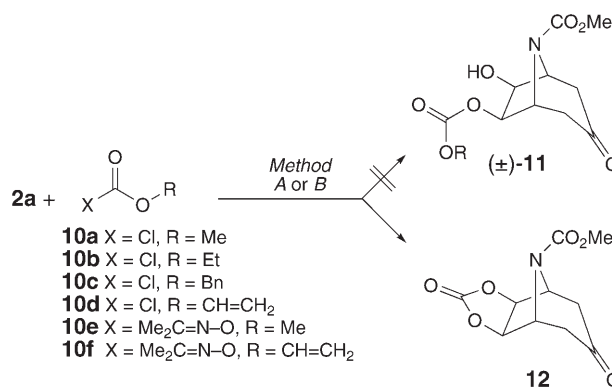
Lipase	Compound	t [h]	Conv. [%]
Novozyme 435	2a	3	57
Novozyme 435	2a	24	93
Chirazyme L-6	2a	24	45
Novozyme 435	(±)- 5d	3	55 (-)- 6f (35%, 98% ee) (+)- 5d (40%, 89% ee)
Chirazyme L-6	(±)- 5d	26	11

As can be seen, the two lipases differed remarkably in their reactivity. Surprisingly, a closer inspection of the NMR spectra revealed that, independent of the type of lipase, the 2-hydroxy-2-(trifluoromethyl)dioxole derivative **9** was obtained as a 3 : 1 mixture of diastereoisomers instead of the expected α -hydroxy-trifluoroacetate analogue of **3a**. A spectroscopic evidence for compound **9** is the quaternary-C signal at δ 115.8 and 116.7 ppm in the ¹³C-NMR spectrum as compared to the signal of the C=O group in **6f** at δ 157.0 ppm. When **2a** was treated with (CF₃CO)₂O, again dioxole derivative **9** was formed as a 3:1 diastereoisomeric mixture. That means that the α -hydroxy trifluoroacetate has an even more pronounced tendency to form the acetoxonium ion than the corresponding α -hydroxy-acetate **3a**.

To check whether vinyl trifluoroacetate may lead to enantioselective acylation in cases where no acyl migration is possible, the MOM derivative (±)-**5d** was resolved with vinyl trifluoroacetate, and either Chirazyme L-6 or Novozyme 435 (*Scheme 5*).

Chirazyme L-6 gave only low conversion. In contrast, Novozyme 435 yielded (–)-**6f** in 35% with 98% ee, and MOM derivative (+)-**5d** in 40% with 89% ee.

In a final attempt, the enzymatic alkoxy-carbonylation²⁾ of unprotected dihydroxy compound **2a** was investigated because the corresponding carbonate is proposed to be less prone to acyl migration. First, as shown in *Scheme 6*, unprotected **2a** was reacted with various carbonic acid derivatives **10a–10d** in the presence of Et₃N and DMAP, but surprisingly, instead of derivative (±)-**11**, only *meso*-carbonate **12** was formed (*Method A*). In the case of **10a** and **10b**, compound **12** was isolated in 91 and 51% yield, respectively.

Scheme 6^{a)}

^{a)} *Method A*: **10a–10b**, Et₃N, DMAP, CH₂Cl₂. *Method B*: **10e** and **10f**, Novozyme 435 or Chirazyme L-6, THF, 40°.

Upon treatment of **2a** with carbonic acid derivatives **10e** and **10f** in the presence of lipases Novozyme 435 or Chirazyme L-6 (*Method B*), again only compound **12** and no trace of the desired carbonate **11** was observed. Carbonate **12** was independently obtained in 58% yield by reaction of **2a** with *N,N*-carbonyldiimidazole in CH₂Cl₂ at room temperature.

Conclusions. – To overcome the racemization by acyl migration, dihydroxy derivative **2a** was protected with MOMCl prior to the lipase-mediated acetylation. Indeed, this synthetic strategy allowed resolution of the MOM derivative (±)-**5d** with vinyl acetate and lipases Chirazyme L-6 and Novozyme 435 to give enantiomerically pure acetate (–)-**6d** (96–99% ee) and (+)-**5d** (up to 82% ee). Deacetylation of (–)-**6d** with K₂CO₃ in MeOH yielded quantitatively (–)-**5d** with 99% ee, being (*R*)-configured at C(6), as assigned by the *Mosher* method. Novozyme 435 also worked well in the resolution of (±)-**5d** with vinyl trifluoroacetate to afford the corresponding acetate (–)-**6f** (98% ee) and (+)-**5d** (89% ee). Thus, in this manner, enantiomerically pure dihydroxytropane building blocks are conveniently accessible, which can be used for

²⁾ For previous work on regio- and/or enantioselective lipase-catalyzed alkoxy-carbonylation of diols, see [25].

further manipulations. In contrast, when a free OH group adjacent to the acyl function is present, as in removal of MOM in **6d** or enzymatic acylation of **2a** with vinyl trifluoroacetate, acyl migration was favored, leading either to racemic products, or the intermediate acetoxonium ions were trapped as dioxole **9**. Even the use of alkoxycarbonylation reagents **10** did not overcome the strong neighboring-group effect of the second OH group in diol **2a**.

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Experimental Part

1. *General*. Commercial reagents were used without further purification unless otherwise indicated. All solvents were distilled prior to use. Reactions were performed in oven-dried glassware. Flash chromatography (FC): was performed on silica gel 60 (230–400 mesh; *Fluka*). GC: *Hewlett-Packard HP 6890* instrument; column *HP 5TA* (30 m × 0.32 mm); temp. program: 16° min⁻¹ gradient from 80° to 300°; *Finnigan Trace GC 2000 Ultra*, column trifluoroacetyl- γ -cyclodextrine (30 m × 0.25 mm), *Lipodex E*, *Bondex un- β* , *Bondex un- α* , *Bondex un- $\alpha + \beta$* , *Amidex P2210*. M.p.: *Büchi SMP-20*; uncorrected. IR Spectra: *Bruker Vector-22 FT-IR* spectrophotometer; in cm⁻¹. ¹H- and ¹³C-NMR spectra: *Bruker Avance-500* instrument; at 500/125 MHz; δ in ppm, *J* in Hz; signal assignments are based on DEPT and COSY experiments; * denotes signals of the minor rotamer, and # denotes the minor diastereoisomer. MS and ESI-MS: *Finnigan MAT-95*, *Varian MAT-711*, and *Bruker Daltonics micrOTOF_Q*; in *m/z* (rel. %).

2. *Methyl 6-Hydroxy-3-oxo-7-[(trimethylsilyl)oxy]-8-azabicyclo[3.2.1]octane-8-carboxylate ((±)-5a)*. A soln. of TMSCl (92 μ l, 0.716 mmol) in CH₂Cl₂ (1 ml) was added dropwise to a soln. of **2a** (140 mg, 0.651 mmol) and ³Pr₂EtN (215 μ l, 1.30 mmol) in CH₂Cl₂ (6 ml) under N₂, and the mixture was stirred for 16 h. After removal of the solvent and all volatile materials *in vacuo*, the residue was chromatographed (SiO₂; AcOEt/hexane 4:1; *R_f* (AcOEt/hexane 2:1) 0.51) to give ((±)-**5a** (110 mg, 0.383 mmol, 60%). Colorless oil. FT-IR (ATR): 3472*m*, 1702*vs*, 1451*s*, 1394*s*, 1253*s*, 1094*s*, 1009*m*, 886*s*, 632*s*. ¹H-NMR (500 MHz, CDCl₃): 0.19 (*s*, Me₃Si, Me₃Si*); 2.40 (*s*, 1 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 2.44 (*s*, 2 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 2.47 (*s*, 1 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 2.53–2.63, 2.65–2.75 (2*m*, 2 × 2 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 3.45 (*br. s*, OH*); 3.54–3.60 (*br. m*, OH); 3.77 (*s*, MeO, MeO*); 3.91–3.96 (*m*, 2 H, H-C(6), H-C(6)*); 4.02 (*s*, 1 H, H-C(7)); 4.03 (*s*, 1 H, H-C(7)*); 4.20–4.27 (*m*, 1 H, H-C(1)* or H-C(5)*); 4.31–4.40 (*br. m*, 2 H, H-C(1), H-C(5)); 4.43–4.48 (*m*, 1 H, H-C(1)* or H-C(5)*). ¹³C-NMR (125 MHz, CDCl₃): 0.0 (Me₃Si, Me₃Si*); 45.7, 46.0, 46.3 (C(2), C(4), C(2)*, C(4)*); 53.2 (MeO, MeO*); 61.4, 61.5, 61.7 (C(1), C(5), C(1)*, C(5)*); 74.1, 74.7, 75.5 (C(6), C(7), C(6)*, C(7)*); 155.3 (COO, COO*); 171.4 (OCO, OCO*); 206.0 (C(3), C(3)*). CI-MS: 288 (40, [M + H]⁺), 256 (100), 240 (20), 213 (20), 197 (10), 155 (15), 123 (30), 103 (15). Anal. calc. for C₁₂H₂₁NO₃Si (287.39): C 50.15, H 7.37, N 4.87; found: C 50.05, H 7.32, N 4.54.

3. *Methyl 7-[(tert-Butyl)dimethylsilyl]oxy]-6-hydroxy-3-oxo-8-azabicyclo[3.2.1]octane-8-carboxylate ((±)-5b)*. A soln. of TBSCl (14.5 mg, 0.095 mmol) in CH₂Cl₂ (1 ml), was added dropwise to a soln. of **2a** (17 mg, 0.079 mmol), Et₃N (0.1 ml), and DMAP (5 mg) in CH₂Cl₂ (1 ml) at 0°, and the mixture was stirred at r.t. for 5 h. Then CH₂Cl₂ (30 ml) was added, and the mixture washed with 1*N* aq. HCl (2 × 10 ml). The org. layer was successively washed with a sat. soln. of NaHCO₃ and H₂O (10 ml each), dried (Na₂SO₄), and concentrated *in vacuo*. The residue was purified by FC (SiO₂; AcOEt/hexane 1:1; *R_f* 0.44) to give ((±)-**5b** (21 mg, 0.064 mmol, 81%). Colorless solid. M.p. 90°. FT-IR (ATR): 3486*m*, 2958*m*, 2857*m*, 1703*vs*, 1446*s*, 1386*vs*, 1251*s*, 1198*s*, 1080*vs*, 1008*s*, 772*vs*. ¹H-NMR (500 MHz, CDCl₃): 0.12 (*s*, Me₂Si); 0.14 (*s*, Me₂Si*); 0.91 (*s*, Me₃C, Me₃C*); 2.40, 2.44, 2.47, 2.49 (4*s*, 4 × 1 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 2.53–2.64, 2.65–2.75 (2*m*, 2 × 2 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 3.45 (*d*, *J* = 5.4, OH*); 3.58 (*d*, *J* = 5.2, OH); 3.77 (*s*, COOMe, COOMe*); 3.92–3.98 (*m*, 2 H, H-C(6), H-C(6)*); 4.05 (*br. s*, 1 H, H-C(7)*); 4.06 (*s*, 1 H, H-C(7)); 4.23–4.29 (*m*, 1 H, H-C(1)* or

H–C(5)*); 4.31–4.41 (*m*, 2 H, H–C(1), H–C(5)); 4.44–4.51 (*m*, 1 H, H–C(1)* or H–C(5)*). ¹³C-NMR (125 MHz, CDCl₃): –5.2, –4.8 (Me₂Si, Me₂Si*); 18.1 (Me₃C, Me₃C*); 25.6 (Me₃C, Me₃C*); 45.4, 45.7, 45.9 (C(2), C(4), C(2)*, C(4)*); 52.9 (MeO, MeO*); 61.2, 61.5 (C(1), C(5), C(1)*, C(5)*); 74.1, 74.6 (C(6), C(6)*); 75.0, 75.7 (C(7), C(7)*); 155.0 (COO, COO*); 205.6 (C(3), C(3)*). CI-MS: 330.2 (100, [M + H]⁺), 298 (10), 272 (75), 257 (20), 240 (20), 212 (15), 197 (10), 127 (10), 75 (10). Anal. calc. for C₁₅H₂₇NO₅Si (329.47): C 54.68, H 8.26, N 4.25; found: C 54.71, H 8.26, N 4.13.

3.1. *Methyl 6-Hydroxy-3-oxo-7-[(triethylsilyl)oxy]-8-azabicyclo[3.2.1]octane-8-carboxylate ((±)-5c)*. As described above for **5b**, from **2a** (81 mg, 0.38 mmol), Et₃N (140 μl), DMAP (12 mg) in CH₂Cl₂ (3 ml), and TESCl (82 μl, 0.49 mmol); FC with AcOEt/hexane 1:2. Yield: 92% (114 mg, 0.35 mmol). Colorless oil. TLC: *R*_f 0.34. FT-IR (ATR): 3475*m*, 2955*m*, 2912*m*, 2878*m*, 1703*vs*, 1449*s*, 1390*s*, 1189*m*, 1091*vs*, 1005*s*, 729*vs*. ¹H-NMR (500 MHz, CDCl₃): 0.65 (*q*, *J* = 8.0, 12 H, SiCH₂Me, SiCH₂Me*); 0.96 (*t*, *J* = 8.0, 18 H, SiCH₂Me, SiCH₂Me*); 2.40 (*s*, 1 H, H–C(2)* or H–C(4)*); 2.43 (*s*, 2 H, H–C(2), H–C(4)); 2.52–2.62, 2.64–2.75 (2*m*, 2 × 2 H, H–C(2), H–C(4), H–C(2)*, H–C(4)*); 3.51–3.57 (br. *m*, OH*); 3.61–3.69 (br. *m*, OH); 3.76 (*s*, COOMe, COOMe*); 3.92–3.96 (*m*, 2 H, H–C(6), H–C(6)*); 4.02–4.07 (*m*, 2 H, H–C(7), H–C(7)*); 4.20–4.28 (br. *m*, 1 H, H–C(1)* or H–C(5)*); 4.30–4.40 (br. *m*, 2 H, H–C(1), H–C(5)); 4.42–4.50 (br. *m*, 1 H–C(1)* or H–C(5)*). ¹³C-NMR (125 MHz, CDCl₃): 4.5 (SiCH₂Me, SiCH₂Me*); 6.5 (SiCH₂Me, SiCH₂Me*); 45.4, 45.7, 45.8, 46.1 (C(2), C(4), C(2)*, C(4)*); 52.9 (MeO, MeO*); 61.3, 61.5 (C(1), C(5), C(1)*, C(5)*); 74.1, 74.6, 75.3 (C(6), C(7), C(6)*, C(7)*); 155.0 (COO, COO*); 205.7 (C(3), C(3)*). CI-MS: 330.1 (100, [M + H]⁺), 300 (90), 268 (65), 255 (15), 240 (20), 225 (15), 196 (10), 155 (20), 127 (10), 103 (10), 87 (10). Anal. calc. for C₁₅H₂₇NO₅Si (329.47): C 54.68, H 8.26, N 4.25; found: C 54.82, H 8.29, N 4.16.

4. *Methyl 6-Hydroxy-7-(methoxymethoxy)-3-oxo-8-azabicyclo[3.2.1]octane-8-carboxylate ((±)-5d)*. MOMCl (140 μl, 1.86 mmol) was added dropwise to a soln. of **2a** (200 mg, 0.93 mmol), ¹Pr₂EtN (790 μl, 4.65 mmol), and DMAP (12 mg, 0.09 mmol) in anh. CH₂Cl₂ (3 ml) at 0°, and the mixture was stirred at 0° for 0.5 h and for a further 16.5 h at r.t. Then, CH₂Cl₂ (30 ml) was added, and the mixture was washed with a sat. soln. of NH₄Cl (20 ml). The org. layer was dried (Na₂SO₄) and concentrated. The residue was purified by FC (SiO₂; AcOEt/hexane 3:1; *R*_f 0.23) to give (±)-**5d** (116 mg, 0.45 mmol, 48%). Yellowish oil. FT-IR (ATR): 3438*m*, 2954*m*, 2903*m*, 2828*m*, 1698*vs*, 1451*s*, 1398*s*, 1192*m*, 1102*s*, 1040*s*. ¹H-NMR (500 MHz, CDCl₃): 2.44 (*s*, 2 H, H–C(2)*, H–C(4)*); 2.47 (*s*, 2 H, H–C(2), H–C(4)); 2.53–2.75 (*m*, 4 H, H–C(2), H–C(4), H–C(2)*, H–C(4)*); 3.23 (br. *s*, OH*); 3.33 (br. *s*, OH); 3.40 (*s*, CH₂OMe, CH₂OMe*); 3.76 (*s*, COOMe, COOMe*); 3.97 (*s*, 1 H, H–C(6)*); 3.99 (*s*, 1 H, H–C(6)); 4.07–4.10 (*m*, 2 H, H–C(7), H–C(7)*); 4.34–4.59 (br. *m*, 4 H, H–C(1), H–C(5), H–C(1)*, H–C(5)*); 4.70 (*d*, *J* = 6.7, CH₂); 4.73 (*d*, *J* = 6.8, CH₂*). ¹³C-NMR (125 MHz, CDCl₃): 45.2, 45.6, 45.7, 46.0 (C(2), C(4), C(2)*, C(4)*); 52.9 (MeO, MeO*); 56.2 (CH₂OMe, CH₂OMe*); 59.0, 61.3 (C(1), C(5), C(1)*, C(5)*); 74.3 (C(6)*); 74.9 (C(6)); 78.9 (C(7)); 79.7 (C(7)*); 96.9 (CH₂, CH₂*); 154.9 (COO, COO*); 205.3 (C(3), C(3)*). EI-MS: 259.1 (10, M⁺), 214 (20), 197 (25), 155 (85), 127 (30), 87 (15), 59 (20), 45 (100), 28 (20). HR-EI-MS: 259.1056 (M⁺, C₁₁H₁₇NO₆⁺; calc. 259.1056).

5. *Methyl 6-Acetoxy-7-(methoxymethoxy)-3-oxo-8-azabicyclo[3.2.1]octane-8-carboxylate (rac-6d)*. *Typical Procedure*. Ac₂O (20 μl, 0.20 mmol) was added to a soln. of **5d** (30 mg, 116 μmol), DMAP (5 mg, 41.0 μmol), and Et₃N (55 μl, 0.31 mmol) in anh. CH₂Cl₂ (1.5 ml), and the mixture was stirred at r.t. for 0.5 h. Then, CH₂Cl₂ (20 ml) was added, and the mixture was washed successively with 0.1*N* aq. NaOH and brine (10 ml each). The org. layer was dried (Na₂SO₄) and concentrated *in vacuo*. The crude product was purified by FC (SiO₂; AcOEt/hexane 2:1; *R*_f (AcOEt/hexane 3:1) 0.32) to give **6d** (26 mg, 86.3 μmol, 74%). Colorless oil. FT-IR (ATR): 2956*m*, 2900*m*, 1739*s*, 1702*vs*, 1450*s*, 1396*s*, 1234*s*, 1196*m*, 1104*s*, 1042*s*. ¹H-NMR (500 MHz, CDCl₃): 2.09 (*s*, MeCO, MeCO*); 2.48–2.50 (*m*, 2 H, H–C(2)* or H–C(4)*); 2.50–2.52 (*m*, 1 H, H–C(2) or H–C(4)); 2.53 (*s*, 1 H, H–C(2) or H–C(4)); 2.57–2.67, 2.76–2.78 (2*m*, 2 × 2 H, H–C(2), H–C(4), H–C(2)*, H–C(4)*); 3.37 (*s*, CH₂OMe, CH₂OMe*); 3.77 (*s*, COOMe, COOMe*); 4.07, 4.08 (2*s*, 2 × 1 H, H–C(7), H–C(7)*); 4.42–4.61 (br. *m*, 4 H, H–C(1), H–C(5), H–C(1)*, H–C(5)*); 4.57 (*d*, *J* = 11.8, CH₂); 4.59 (*d*, *J* = 11.8, CH₂*); 4.93 (*s*, 1 H, H–C(6)); 4.94 (*s*, 1 H, H–C(6)*). ¹³C-NMR (125 MHz, CDCl₃): 20.6 (MeCO, MeCO*); 45.1, 45.6, 45.8, 46.2 (C(2), C(4), C(2)*, C(4)*); 53.0 (COOMe, COOMe*); 56.1 (CH₂OMe, CH₂OMe*); 58.0, 59.4 (C(1), C(5), C(1)*, C(5)*); 76.2 (C(6), C(6)*); 78.9, 79.8 (C(7), C(7)*); 97.3 (CH₂, CH₂*); 154.5 (COO, COO*); 170.3

(OCO, OCO*); 204.6 (C(3), C(3)*). EI-MS: 301.1 (5, M^{+}), 256 (20), 241 (25), 214 (20), 196 (60), 154 (100), 127 (20), 59 (15), 45 (100), 28 (15). HR-EI-MS: 301.1161 (M^{+} ; $C_{13}H_{19}NO_7^{+}$; calc. 301.1162).

5.1. *Methyl 6-Acetoxy-7-[(tert-butyl)dimethylsilyloxy]-3-oxo-8-azabicyclo[3.2.1]octane-8-carboxylate (rac-6b)*. R_f (AcOEt/hexane 1:3) 0.25. Yield: 91%. Colorless oil. FT-IR (ATR): 2955m, 2930m, 2857m, 1740s, 1703vs, 1449s, 1388s, 1237vs, 1194s, 1101s, 1007s, 775s, 679m. 1H -NMR (500 MHz, $CDCl_3$): 0.03 (s, Me_2Si); 0.08 (s, Me_2Si^*); 0.87 (s, Me_3C , Me_3C^*); 2.08 (s, MeCO); 2.09 (s, MeCO*); 2.43, 2.46 (2s, 2×1 H, H-C(2), H-C(2)*, H-C(4), H-C(4)*); 2.49, 2.53 (2d, $J=10.5$, 2×1 H, H-C(2), H-C(2)*, H-C(4), H-C(4)*); 2.59 (t, $J=5.0$, 2 H, H-C(2), H-C(2)*, H-C(4), H-C(4)*); 2.68–2.78 (m, 2 H, H-C(2), H-C(2)*, H-C(4), H-C(4)*); 3.77 (s, COOMe); 3.78 (s, COOMe*); 4.17 (d, $J=6.2$, 1 H, H-C(7)*); 4.19 (d, $J=6.0$, 1 H, H-C(7)); 4.27–4.31, 4.33–4.37, 4.44–4.48, 4.56–4.60 (4m, 4×1 H, H-C(1), H-C(1)*, H-C(5), H-C(5)*); 4.86 (d, $J=5.8$, 1 H, H-C(6)); 4.89 (d, $J=5.8$, 1 H, H-C(6)*). ^{13}C -NMR (125 MHz, $CDCl_3$): –5.3, –5.1 (Me_2Si , Me_2Si^*); 18.1 (Me_3C , Me_3C^*); 20.7 (MeCO, MeCO*); 25.6 (Me_3C , Me_3C^*); 45.3, 45.6, 45.9, 46.1 (C(2), C(4), C(2)*, C(4)*); 52.9 (MeO, MeO*); 58.3, 58.4, 61.8 (C(1), C(5), C(1)*, C(5)*); 75.2, 75.9 (C(7), C(7)*); 76.5, 77.1 (C(6), C(6)*); 154.5, 154.7 (COO, COO*); 170.1 (OCO, OCO*); 205.0 (C(3), C(3)*). CI-MS: 372.2 (100, $[M+H]^+$), 340 (10), 314 (50), 272 (15), 240 (10), 212 (10), 155 (5), 127 (5), 43 (5). HR-ESI-MS: 394.1661 ($[M+Na]^+$, $C_{17}H_{29}NNaO_6Si^+$; calc. 394.1662).

5.2. *Methyl 6-Acetoxy-3-oxo-7-[(triethylsilyloxy)-8-azabicyclo[3.2.1]octane-8-carboxylate (rac-6c)*. R_f (AcOEt/hexane 1:3) 0.21. Yield: 78%. Colorless oil. FT-IR (ATR): 2955m, 2912m, 2878m, 1740s, 1703vs, 1449s, 1386s, 1229vs, 1102s, 1005s, 727s. 1H -NMR (500 MHz, $CDCl_3$): 0.59 (q, $J=8.0$, 12 H, $SiCH_2Me$, $SiCH_2Me^*$); 0.94 (t, $J=8.0$, 18 H, $SiCH_2Me$, $SiCH_2Me^*$); 2.08 (s, MeCO); 2.09 (s, MeCO*); 2.43, 2.46 (2s, 2×1 H, H-C(2), H-C(2)*); 2.47–2.51, 2.51–2.55 (2m, 2×1 H, H-C(4), H-C(4)*); 2.56–2.65, 2.68–2.78 (2m, 2×2 H, H-C(2), H-C(2)*, H-C(4), H-C(4)*); 3.77 (s, MeO, MeO*); 4.16–4.22 (m, 2 H, H-C(7), H-C(7)*); 4.23–4.29, 4.31–4.36, 4.42–4.48, 4.53–4.59 (4m, 4×1 H, H-C(1), H-C(5), H-C(1)*, H-C(5)*); 4.84–4.90 (m, 2 H, H-C(6), H-C(6)*). ^{13}C -NMR (125 MHz, $CDCl_3$): 4.3 ($SiCH_2Me$, $SiCH_2Me^*$); 6.6 ($SiCH_2Me$, $SiCH_2Me^*$); 20.7 (MeCO, MeCO*); 45.3, 45.7, 45.8, 46.1 (C(2), C(4), C(2)*, C(4)*); 52.9 (MeO, MeO*); 58.3, 61.8 (C(1), C(5), C(1)*, C(5)*); 74.8 (C(7)*); 75.6 (C(7)); 76.5 (C(6)); 77.1 (C(6)*); 154.7 (COO, COO*); 170.2 (OCO, OCO*); 205.0 (C(3), C(3)*). CI-MS: 372.1 (100, $[M+H]^+$), 342 (75), 330 (30), 300 (60), 268 (25), 237 (15), 196 (5), 145 (10), 49 (10). HR-ESI-MS: 394.1658 ($[M+Na]^+$, $C_{17}H_{29}NNaO_6Si^+$; calc. 394.1662).

5.3. *Methyl 7-(Methoxymethoxy)-3-oxo-6-(trifluoroacetoxy)-8-azabicyclo[3.2.1]octane-8-carboxylate (rac-6f)*. $(CF_3CO)_2O$ instead of Ac_2O . R_f (AcOEt/hexane 1:2) 0.15. Yield: 53%. Colorless oil. FT-IR (ATR): 3392w, 2958w, 1784w, 1696vs, 1457s, 1402s, 1193s, 1107m, 632s. 1H -NMR (500 MHz, $CDCl_3$): 2.51, 2.52, 2.54, 2.55 (4s, 4×1 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 2.68, 2.79 (2dt, $J=17.0$, 4.5, 2×2 H, H-C(2), H-C(4), H-C(2)*, H-C(4)*); 3.36 (s, CH_2OMe , CH_2OMe^*); 3.80 (s, COOMe, COOMe*); 4.14, 4.15 (2s, 2×1 H, H-C(7), H-C(7)*); 4.47–4.52 (br. m, 1 H, H-C(1) or H-C(5)); 4.54 (d, $J=7.0$, CH_2); 4.58 (d, $J=7.0$, CH_2); 4.57–4.62, 4.64–4.69, 4.73–4.78 (3m, 3×1 H, H-C(1) or H-C(5), H-C(1)*, H-C(5)*); 5.06–5.12 (m, 2 H, H-C(6), H-C(6)*). ^{13}C -NMR (125 MHz, $CDCl_3$): 45.2, 45.4, 45.7, 45.9 (C(2), C(4), C(2)*, C(4)*); 53.2 (COOMe, COOMe*); 56.4 (CH_2OMe , CH_2OMe^*); 57.5, 59.0, 59.1 (C(1), C(5), C(1)*, C(5)*); 78.5, 79.3 (C(7), C(7)*); 79.1, 79.9 (C(6), C(6)*); 97.2, 97.3 (CH_2 , CH_2^*); 114.4 (q, $J=285.2$, CF_3 , CF_3^*); 154.4 (COO, COO*); 157.0 (q, $J=36.9$, $OCOCF_3$, $OCOCF_3^*$); 204.0 (C(3), C(3)*). CI-MS: 356.1 (20, $[M+H]^+$), 324 (100), 294 (50), 260 (15), 228 (45), 196 (35), 154 (20), 115 (60), 45 (35). Anal. calc. for $C_{13}H_{16}F_3NO_7$ (355.27): C 43.95, H 4.54, N 3.94; found: C 43.98, H 4.69, N 3.85.

5.4. *Methyl 4,5,6,7,8,8a-Hexahydro-2-hydroxy-6-oxo-2-(trifluoromethyl)-4,8-epimino-3aH-cyclohepta[d][1,3]dioxole-9-carboxylate (9)*. $(CF_3CO)_2O$ instead of Ac_2O . R_f (AcOEt) 0.61. Yield: 73% (82% GC purity). Colorless solid. M.p. 183°. FT-IR (ATR): 3301m, 1694vs, 1674vs, 1448s, 1402vs, 1228s, 1173vs, 1056vs, 984vs, 644s. 1H -NMR (500 MHz, $(CD_3)_2CO$): 2.43–2.49 (m, 4 H, H-C(2), H-C(4), H-C(2)[#], H-C(4)[#]); 2.70–2.76 (m, 4 H, H-C(2), H-C(4), H-C(2)[#], H-C(4)[#]); 3.70 (s, Me[#]); 3.72 (s, Me); 4.48 (br. d, $J=5.3$, 2 H, H-C(1), H-C(5)); 4.53, 4.57 (2d, $J=5.2$, 2×1 H, H-C(1)[#], H-C(5)[#]); 4.77 (d, $J=3.8$, 2 H, H-C(6), H-C(7)); 4.93 (d, $J=1.6$, 2 H, H-C(6)[#], H-C(7)[#]); 7.57 (br. s, OH[#]); 8.24 (br. s, OH). ^{13}C -NMR (500 MHz, $(CD_3)_2CO$): 43.7, 44.2 (C(2), C(4)); 44.2, 44.7 (C(2)[#], C(4)[#]); 52.4 (MeO[#]); 52.6 (MeO); 58.7, 58.8 (C(1)[#], C(5)[#]); 59.3 (C(1), C(5)); 84.2, 85.1 (C(6), C(7)); 84.3, 85.0 (C(6)[#], C(7)[#]);

115.8 ($q, J = 25.4, F_3COH^{\#}$); 116.7 ($q, J = 25.3, F_3COH$); 120.5 ($q, J = 285.1, F_3C^{\#}$); 121.5 ($q, J = 288.3, F_3C$), 154.5 ($CO^{\#}$); 155.1 (CO); 203.9 ($C(3)$); 204.3 ($C(3)^{\#}$). EI-MS: 311.1 (35, $M^{+\cdot}$), 242 (10), 197 (10), 154 (100), 127 (30), 97 (10), 82 (10), 59 (15), 42 (10). Anal. calc. for $C_{11}H_{12}F_3NO_6$ (311.21): C 42.45, H 3.89, N 4.50; found: C 42.49, H 4.06, N 4.31.

6. *General Procedure for Enzymatic Acylations.* To a soln. of **2a** or *rac-5d* (0.07 mmol) in the given solvent (1 ml) was added vinyl acetate or vinyl trifluoroacetate (13 μ l), molecular sieves (4 Å; 4 pellets), and the respective enzyme (20–27 mg). The mixture was stirred at 40°. At time intervals from 1 to 3 h and sedimentation of the enzyme, aliquots of 30 μ l were taken from the supernatant, filtered, diluted with CH_2Cl_2 (300 μ l), and directly analyzed by cap. GC on chiral phase *Bondex un- β* . The reaction was terminated after 48 h.

For separation and isolation of (–)-**6d** or (–)-**6f**, and (+)-**5d**, the mixture was filtered through *Celite*, and the filtrate was concentrated. Chromatography of the residue on SiO_2 with AcOEt/hexane 2 : 1 gave in a first fraction **6d** (R_f (AcOEt/hexane 3 : 1) 0.32) and in a second fraction **5d** (R_f (AcOEt/hexane 3 : 1) 0.23). Second chromatography on SiO_2 with AcOEt/hexane 1 : 2.5 \rightarrow 1 : 2 gave in a first fraction **6f** (R_f (AcOEt/hexane 3 : 1) 0.63) and in a second fraction **5d** (R_f (AcOEt/hexane 3 : 1) 0.23).

7.1. *Methyl (6R,7S)-7-(Methoxymethoxy)-6-[(R)- α -methoxy- α -(trifluoromethyl)phenylacetyl]-3-oxo-8-azabicyclo[3.2.1]octane-8-carboxylate ((6R,7S)-(R)-**6e**).* To a soln. of (–)-**5d** (17 mg, 65.6 μ mol) in anh. pyridine (0.5 ml) was added (+)-(*S*)-MTPACl (24 μ l), and the mixture was stirred at r.t. for 18 h. After addition of toluene (1 ml), the mixture was concentrated three times *in vacuo*. The residue was purified by FC (SiO_2 ; AcOEt/hexane 1 : 1; R_f 0.34) to give (6R,7S)-(R)-**6e** (24 mg, 50.5 μ mol, 77 %). Colorless solid. M.p. 105°. FT-IR (ATR): 2963 w , 2860 w , 1742 vs , 1703 vs , 1451 s , 1393 s , 1337 s , 1260 vs , 1158 vs , 1105 s , 1045 vs , 996 vs , 921 s , 803 m , 771 s , 715 vs , 696 vs , 640 s . 1H -NMR (500 MHz, $C_2D_2Cl_4$, 393 K): 2.46 ($dt, J = 16.8, 1.7, H_{eq}-C(4)$); 2.47 ($dt, J = 16.8, 1.7, H_{eq}-C(2)$); 2.65 ($dd, J = 16.8, 5.3, H_{ax}-C(2)$); 2.67 ($dd, J = 16.8, 5.3, H_{ax}-C(4)$); 3.30 (s, CH_2OMe); 3.54 (s, MeO); 3.70 ($s, COOMe$); 4.14 ($d, J = 6.2, H-C(7)$); 4.50 (s, CH_2O); 4.52–4.57 ($m, H-C(1), H-C(5)$); 5.08 ($d, J = 6.2, 1 H, H-C(6)$); 7.33–7.37 ($m, 3 arom. H$); 7.56–7.58 ($m, 2 arom. H$). ^{13}C -NMR (125 MHz, $C_2D_2Cl_4$, 393 K): 44.9 ($C(4)$); 45.4 ($C(2)$); 52.1 ($COOMe$); 54.7 (MeO); 55.3 (CH_2OMe); 57.9, 58.6 ($C(1), C(5)$); 78.3 ($C(6)$); 79.3 ($C(7)$); 96.4 (CH_2O); 127.1, 127.6, 128.9, 131.7 (Ph); 153.8 (NCO); 165.4 (CO); 202.9 ($C(3)$). EI-MS: 475.1 (10, $M^{+\cdot}$), 430 (30), 370 (5), 241 (25), 210 (10), 197 (35), 189 (100), 154 (50), 127 (10), 114 (15), 105 (10), 45 (50). HR-ESI-MS: 498.1330 ($[M + Na]^+$, $C_{21}H_{24}F_3NNaO_6^{\ddagger}$; calc. 498.1352).

7.2. *Methyl (6R,7S)-7-(Methoxymethoxy)-6-[(S)- α -methoxy- α -(trifluoromethyl)phenylacetyl]-3-oxo-8-azabicyclo[3.2.1]octane-8-carboxylate ((6R,7S)-(S)-**6e**).* As described above, from (–)-**5d** and (–)-(*R*)-MTPACl. R_f (AcOEt/hexane 1 : 1) 0.34. Yield: 40%. Colorless solid. 1H -NMR (500 MHz, $C_2D_2Cl_4$, 393 K): 2.46 ($dt, J = 16.8, 1.7, H_{eq}-C(2)$); 2.51 ($dt, J = 16.8, 1.7, H_{eq}-C(4)$); 2.66 ($dd, J = 16.8, 5.3, H_{ax}-C(2)$); 2.67 ($dd, J = 16.8, 5.3, H_{ax}-C(4)$); 3.25 (s, CH_2OMe); 3.50 (s, MeO); 3.71 ($s, COOMe$); 4.11 ($d, J = 6.2, H-C(7)$); 4.40 ($d, J = 12.1, CH_2O$); 4.42 ($d, J = 12.1, CH_2O$); 4.52–4.55 ($m, H-C(1)$); 4.58–4.61 ($m, H-C(5)$); 5.10 ($d, J = 6.2, H-C(6)$); 7.33–7.39 ($m, 3 arom. H$); 7.54–7.57 ($m, 2 arom. H$). ^{13}C -NMR (125 MHz, $C_2D_2Cl_4$, 393 K): 44.9 ($C(4)$); 45.5 ($C(2)$); 52.1 ($COOMe$); 54.6 (MeO); 55.2 (CH_2OMe); 58.0 ($C(5)$); 58.6 ($C(1)$); 78.0 ($C(6)$); 79.1 ($C(7)$); 84.4 ($q, J = 28.4, CCF_3$); 96.2 (CH_2O); 122.7 ($q, J = 289.3, CF_3$); 127.2, 127.6, 128.9, 131.6 (Ph); 153.8 (NCO); 165.4 (CO); 202.9 ($C(3)$).

8. *Methyl 4,5,6,7,8,8a-Hexahydro-2,6-dioxo-4,8-epimino-3aH-cyclohepta[d][1,3]dioxole-9-carboxylate (12).* To a soln. of **2a** (50 mg, 0.233 mmol), DMAP (2 mg, 16.4 μ mol), and Et_3N (90 μ l, 0.652 mmol) in anh. CH_2Cl_2 (1.5 ml) was added at 0° methyl chloroformate (18 μ l, 0.233 mmol), and the mixture was stirred at r.t. for 3 h. Then, CH_2Cl_2 (10 ml) was added, and the mixture was successively washed with a soln. of $NaHCO_3$ and brine (10 ml each). The org. layer was dried (Na_2SO_4) and concentrated *in vacuo*. The crude product was purified by FC (SiO_2 ; AcOEt/hexane 2 : 1; R_f (AcOEt/hexane 3 : 1) 0.40) to give **12** (51 mg, 0.212 mmol, 91 %). Colorless solid. M.p. 159°. FT-IR (ATR): 3011 w , 2962 w , 2908 w , 1779 vs , 1713 vs , 1697 vs , 1453 s , 1379 vs , 1166 vs , 1078 vs , 1053 vs , 979 s , 766 vs , 687 s . 1H -NMR (300 MHz, $CDCl_3$): 2.52 ($br. d, J = 16.8, 2 H, H-C(2), H-C(4)$); 2.68–2.94 ($br. m, 2 H, H-C(2), H-C(4)$); 3.83 (s, Me); 4.61–4.82 ($br. m, H-C(1), H-C(5)$); 4.90 ($s, H-C(6), H-C(7)$). ^{13}C -NMR (125 MHz, $CDCl_3$): 44.0, 44.5 ($C(2), C(4)$); 53.6 (Me); 58.6 ($C(1), C(5)$); 81.0, 81.6 ($C(6), C(7)$); 153.1 (OCO); 154.4 (COO); 202.7 ($C(3)$). EI-MS: 241.1 (30, $M^{+\cdot}$), 154 (100), 127 (55), 82 (20), 59 (30), 42 (30). HR-ESI-MS: 264.0474 ($[M + Na]^+$, $C_{10}H_{11}NNaO_6^{\ddagger}$; calc. 264.0484).

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